

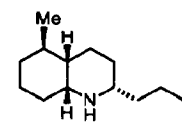
Total Synthesis of (–)-Pumiliotoxin C by Aqueous Intramolecular Acylnitroso Diels–Alder Approach

Masaichi Naruse, Sakae Aoyagi, and Chihiro Kibayashi*

Tokyo College of Pharmacy, Horinouchi, Hachioji, Tokyo 192-03, Japan

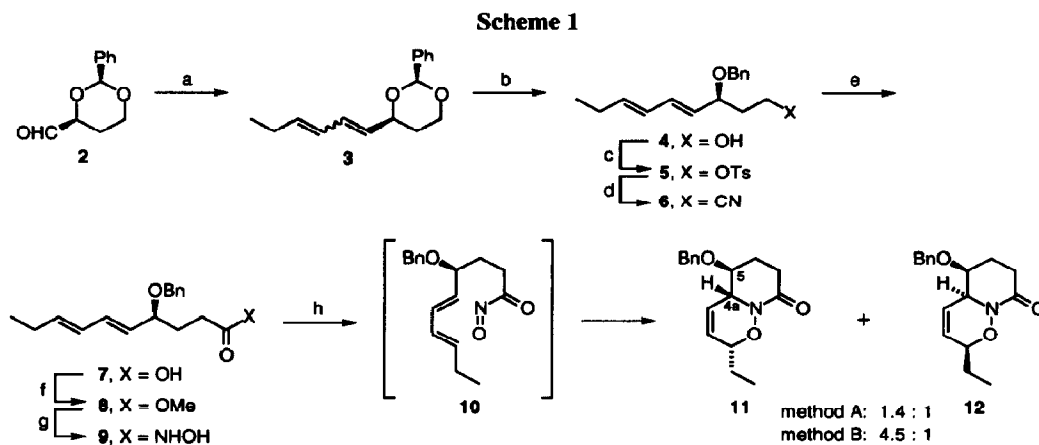
Abstract: A chiral approach to (–)-pumiliotoxin C via a stereoselective intramolecular hetero Diels–Alder reaction of a chiral acylnitroso compound performed in an aqueous medium is described.

Recently we reported that, in aqueous media, intramolecular hetero Diels–Alder cycloadditions of the chiral acylnitroso compounds show a pronounced enhancement of the diastereoselectivity compared with the reaction under conventional nonaqueous conditions,¹ and described an application of this reaction to an enantioselective synthesis of (–)-swainsonine.² Our interest in this area led us to investigate a further application of this strategy for the chiral preparation of natural products, and we now wish to report the enantioselective synthesis of (–)-pumiliotoxin C (decahydroquinoline *cis*-195A) (**1**),^{3,4} isolated from skin extracts of the Panamanian poison-frog *Dendrobates pumilio*⁵ as the first member of one major class of dendrobatid alkaloids.⁶

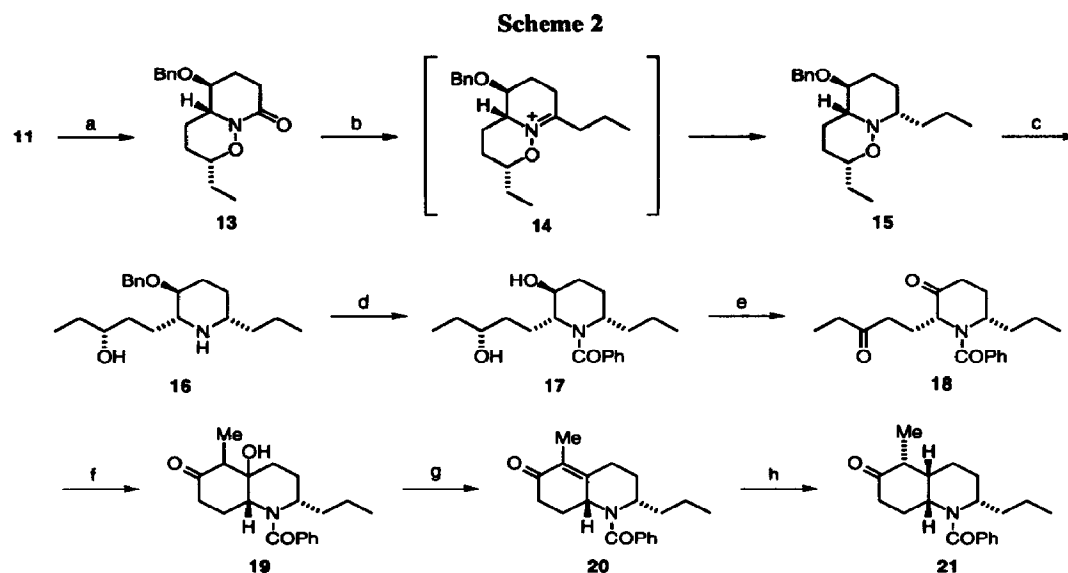


(–)-Pumiliotoxin C (**1**)

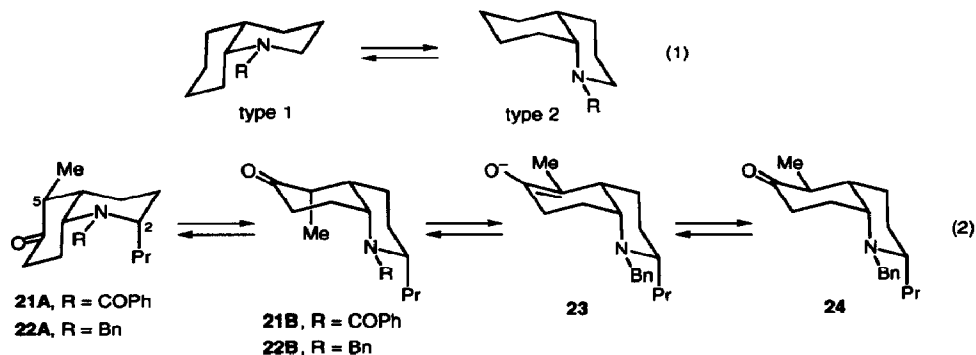
Wittig reaction of (2*S*,4*S*)-4-formyl-2-phenyl-1,3-dioxane (**2**), available from L-malic acid,⁷ with *trans*-2-pentenyltriphenylphosphonium bromide (*t*-BuOK, THF) proceeded without epimerization to give a mixture of (*E*)- and (*Z*)-dienes **3** in 95% combined yield (Scheme 1). Ring-opening of the benzylidene acetal with DIBALH gave an *E/Z* mixture of the alcohol **4** (86%), which underwent photoisomerization in benzene in the presence of iodine to generate the geometrically pure (4*E*)-diene **4** in 69% overall yield. This compound was converted to the ester **8** by standard procedures involving sequential tosylation, cyanation, alkaline hydrolysis, and then diazomethane esterification. Treatment of the ester **8** with hydroxylamine under basic conditions (methanolic KOH) afforded the hydroxamic acid **9** (89%). Subsequent periodate oxidation of **9** under nonaqueous conditions (method A: Pr₄NIO₄, CHCl₃, 0 °C) resulted in in situ generation of the intermediacy acylnitroso diene **10**, which spontaneously underwent intramolecular cycloaddition to produce poor diastereoselectivity (1.4:1) of the *trans* (with respect to 4a-H and 5-H) vs. *cis* adducts (**11** and **12**). However, the use of an aqueous solvent system (method B: H₂O–MeOH, 6:1) instead of CHCl₃ significantly enhanced the *trans* isomer ratio up to 4.5:1.



After catalytic hydrogenation of the olefin moiety of the *trans* adduct **11**, introduction of the propyl side chain to the C-8 position of **13** was achieved in a completely stereoselective manner according to the two-step process previously reported from this laboratory.⁸ Thus, **13** was subjected to Grignard reaction with PrMgBr followed by reduction of the resulting iminium salt **14** in an acidic medium (AcOH, THF) with NaBH_3CN to provide **15** as a single diastereomer in 96% overall yield from **11**. Reductive N–O bond cleavage with Zn and aqueous AcOH followed by N-benzylation of the resulting amino alcohol **16** gave **17** (89%). After hydrogenolysis over palladium on carbon, the resulting diol **17** was converted to the diketone **18** by PCC oxidation. Aldol cyclization of **18** (methanolic KOH, $0\text{ }^\circ\text{C}$) followed by dehydration (KHSO_4 , MeOH, r.t.) yielded the bicyclic enone **20**. Subsequent catalytic hydrogenation of the C-4a–C-5 double bond exclusively occurred from the less hindered β -face to establish the *cis*-fused decahydroquinoline ring system **21**. In this transformation, however, the resulting methyl substitution at C-5 was of the incorrect α -configuration, which relates to 5-epipumiliotoxin C.⁹ Booth¹⁰ reported that conformationally mobile *cis*-decahydroquinoline itself has a strong preference for type 2 conformation, whereas the *N*-acyl derivatives prefer type 1 to avoid the $A^{1,3}$ type strain between the *N*-substituent and the C-8 methylene group (eq 1). This should hold even more so for the *N*-benzoyl-*cis*-decahydroquinolone **21**, which should exist preferentially in the conformation **21A** (type 1) with the 2-propyl and 5-methyl groups occupying pseudo-equatorial and axial positions, respectively (eq 2). The alternative conformation **21B** (type 2) is unlikely owing to the additional $A^{1,3}$ type strain between the *N*-benzoyl group and the propyl group at C-2. However, when the *N*-benzoyl group is replaced by the *N*-benzyl group as in **22**, it could lead to the absence of such allylic strain and should prefer the conformation **22B** (type 2) with the 2-propyl group equatorial, which avoids the 1,3-diaxial interaction between the axial 2-propyl group and the C-8 ring methylene group. However, a steric interaction due to the pseudo-axial methyl group at C-5 is present in **22B**; hence, upon isomerization through its enolate **23**, **22B** would as shown in eq 2 be converted to thermodynamically more stable **24** in which the C-5 methyl substitution is equatorial with the desired β -configuration.



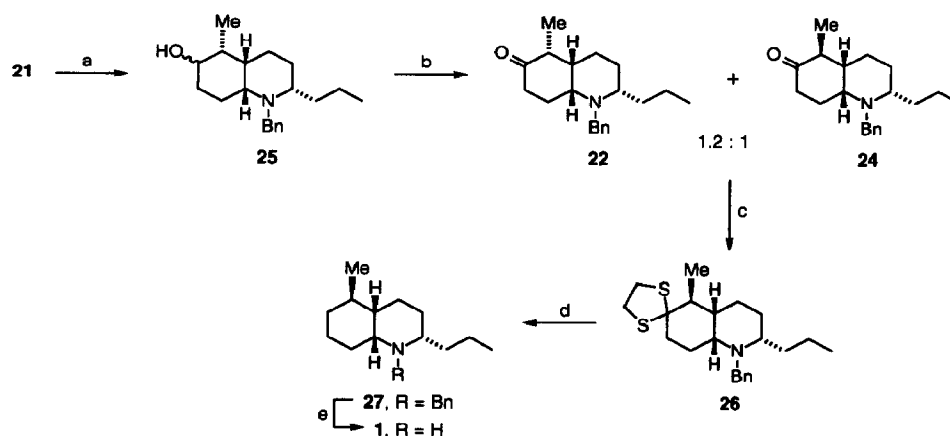
(a) H_2 , 10% Pd-C, THF, r.t. (100%); (b) PrMgBr , THF, 0 °C, then NaBH_3CN , AcOH, THF, 0 °C (96%); (c) Zn , 85% aq. AcOH, 60 °C (91%); (d) i, PhCOCl , CH_2Cl_2 10% aq. K_2CO_3 , then KOH (98%); ii, H_2 , 10% Pd-C, MeOH, r.t. (99%); (e) PCC , CH_2Cl_2 , r.t. (74%); (f) KOH , MeOH, 0 °C (74% or 85% based on starting material recovered); (g) KHSO_4 , MeOH, r.t. (50% or 72% based on starting material recovered); (h) H_2 , 10% Pd-C, 1 N HCl, r.t. (71%).



With these considerations in mind, our attention next focused on epimeric transformation of **22** with the benzyl group as a nitrogen protecting group. Thus, the benzamide **21** was converted by reduction with LiAlH_4 to the *N*-benzyl derivative **25** as a mixture of the alcohol epimers (87%). This mixture was oxidized under Swern conditions to afford a 1.2:1 (by ^1H NMR) epimeric mixture of the 5 α - and 5 β -methyl isomers **22** and **24**, respectively. The formation of the 5 β isomer **24** in this reaction is consistent with concomitant epimerization of the initially formed **22** under the reaction conditions. After prolonged heating (4 days) of this epimeric mixture of **22** and **24** with $\text{Zn}(\text{OTf})_2$ in CH_2Cl_2 , the reaction mixture was treated with 1,2-ethanedithiol (reflux, 4 days). Upon this one-pot procedure, epimerization at C-5 as expected followed by dithioketalization proceeded to produce the thermodynamically more stable 5 β -methyl epimer **26** as a single

isomer (based on 400-MHz ^1H NMR) in 66% yield. Finally, **26** with the correctly oriented side chains at C-2 and C-5 underwent desulfurization with Raney Ni followed by hydrogenolytic removal of the benzyl protecting group to yield (–)-pumiliotoxin C (**1**). Treatment with methanolic HCl and subsequent recrystallization from *i*-PrOH–Et₂O gave the HCl salt of **1** as colorless needles, mp 284–288 °C (sealed capillary) [lit.^{3c} mp 286–288 °C (sealed capillary under Ar)]; $[\alpha]^{24}_{\text{D}} -15.2^\circ$ (*c* 0.46, MeOH) [lit.^{3c} $[\alpha]^{21}_{\text{D}} -16.2^\circ$ (*c* 1.00, MeOH)]. The synthetic sample of **1** exhibited ^1H and ^{13}C NMR spectra identical with those reported^{5b} for natural pumiliotoxin C.

Scheme 3



(a) LiAlH_4 , THF, reflux (87%); (b) $(\text{COCl})_2$, DMSO, CH_2Cl_2 , -90°C , then Et_3N , r.t. (84%); (c) $\text{Zn}(\text{OTf})_2$, CH_2Cl_2 , reflux, 4 days, then $(\text{CH}_2\text{SH})_2$, reflux, 4 days (66%); (d) Raney Ni, dioxane, reflux (68%); (e) H_2 , 10% Pd–C, MeOH (92%).

References and Notes

- Naruse, M.; Aoyagi, S.; Kibayashi, C. *Tetrahedron Lett.* **1994**, *35*, 595.
- Naruse, M.; Aoyagi, S.; Kibayashi, C. *J. Org. Chem.* **1994**, *59*, 1358.
- For chiral syntheses of natural (–)-pumiliotoxin C, see: (a) Oppolzer, W. Flakamp, E. *Helv. Chim. Acta* **1977**, *60*, 204. (b) Bonin, M.; Royer, J.; Grierson, D. S.; Husson, H.-P. *Tetrahedron Lett.* **1986**, *27*, 1569. (c) Murahashi, S.; Sasao, S.; Saito, E.; Naota, T. *J. Org. Chem.* **1992**, *57*, 2521. Murahashi, S.; Sasao, S.; Saito, E.; Naota, T. *Tetrahedron* **1993**, *49*, 8805. (d) Comins, D. L.; Dehghani, A. *J. Chem. Soc., Chem. Commun.* **1993**, 1838. (e) The unnatural (+)-enantiomer: Schultz, A. G.; McCloskey, P. J.; Court, J. J. *J. Am. Chem. Soc.* **1987**, *109*, 6493. See also ref 3a.
- For recent syntheses of racemic pumiliotoxin C, see: (a) Comins, D. L.; Dehghani, A. *Tetrahedron Lett.* **1991**, *32*, 5697. (b) Polnieszek, R. P.; Dillard, L. D. *J. Org. Chem.* **1992**, *57*, 4110. (c) Brandi, A.; Cordero, F. M.; Goti, A.; Guarna, A. *Tetrahedron Lett.* **1992**, *33*, 6697. (d) Meyers, A. I.; Milot, G. *J. Am. Chem. Soc.* **1993**, *115*, 6652.
- (a) Daly, J. W.; Tokuyama, T.; Habermehl, G.; Karle, I. L.; Witkop, B. *Liebigs Ann. Chem.* **1969**, 729, 198. (b) Tokuyama, T.; Tsujita, T.; Shimada, A.; Garraffo, H. M.; Spande, T. F.; Daly, J. W. *Tetrahedron* **1991**, *47*, 5401.
- Daly, J. W.; Spande, T. F. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Wiley-Interscience: New York, 1986; Vol. 4, pp 1–274.
- Corcoran, R. C. *Tetrahedron Lett.* **1990**, *31*, 2101. Thiam, M.; Slassi, A.; Chastrette, F.; Amouroux, R. *Synth. Commun.* **1992**, *22*, 83.
- Watanabe, Y.; Iida, H.; Kibayashi, C. *J. Org. Chem.* **1989**, *54*, 4088.
- Paulvannan, K.; Stille, J. R. *Tetrahedron Lett.* **1993**, *34*, 6673.
- Booth, H.; Bostock, A. H. *J. Chem. Soc., Perkin Trans. 2* **1972**, 615.

(Received in Japan 11 July 1994; accepted 2 September 1994)